

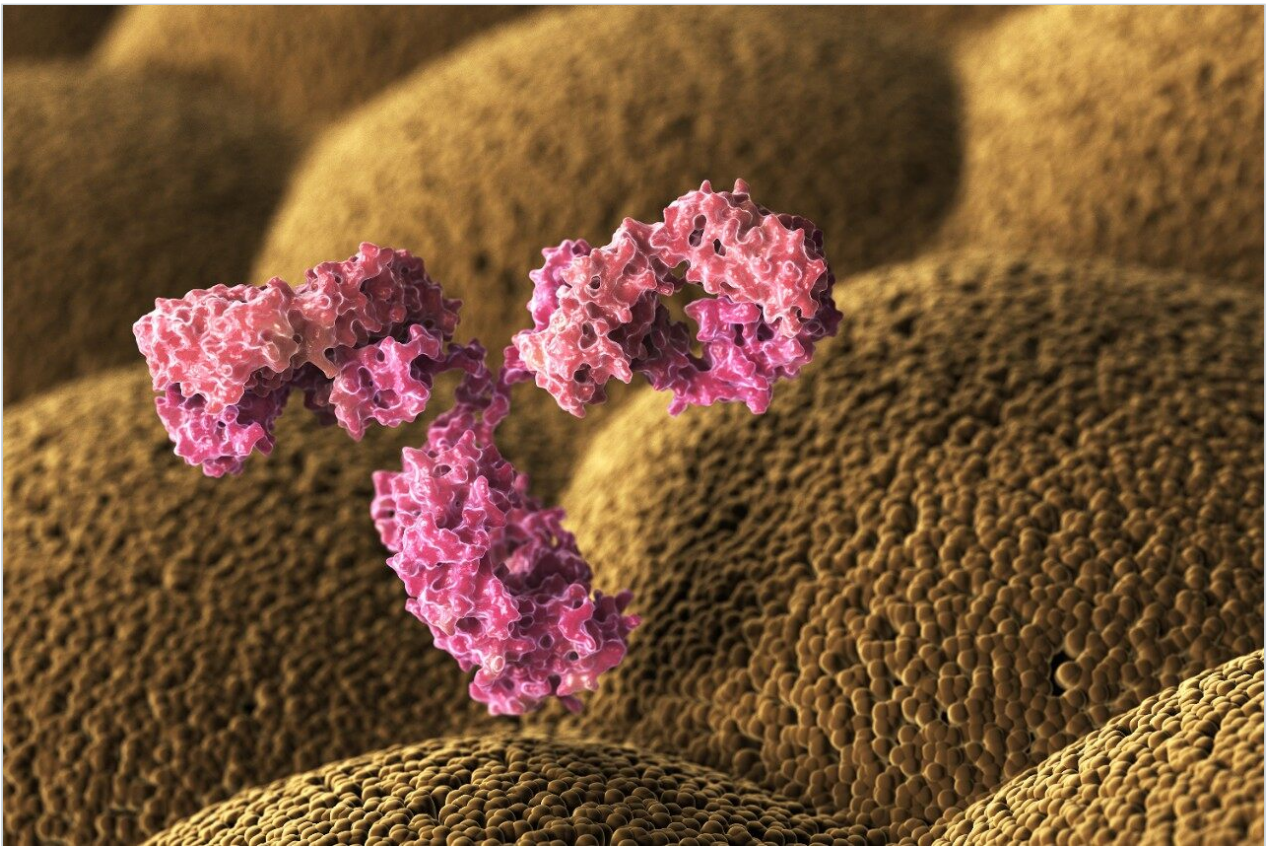
应用纪要

## Comparing Mobile Phase Additives for the Separation of mAb Tryptic Peptides: A Case Study on Formic, Difluoroacetic, and Trifluoroacetic Acid

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## Abstract

This application note expands upon the noted advantages of DFA. With this work, we have explored the use of purified DFA (IonHance DFA) for peptide mapping and found value in its unique selectivity and ability to afford low adduct mass spectra. Moreover, we have compared the performance of DFA peptide mapping against traditional approaches based on FA and TFA.

### Benefits

- Alternative chromatographic selectivity with DFA versus FA or TFA
- High peak capacity separations
- Improved MS sensitivities versus TFA
- MS-grade purity to facilitate the acquisition of high-quality mass spectra

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## Introduction

Peptide mapping is an important technique for the characterization of biopharmaceuticals. Peptide mapping can report on the primary structure of a protein through an examination of its amino acid sequence. The separations underlying this approach are based upon reversed-phase liquid chromatography and are readily coupled to mass spectrometry (RP LC-MS). Separations using spectroscopic detection often employ trifluoroacetic acid (TFA) in the mobile phase, a strong, hydrophobic acid and an effective ion pair for minimizing chromatographic secondary interactions. Conversely, to avoid ion suppression, LC-MS analyses are often performed with a weaker ion pairing modifier, like formic acid (FA), but with a compromise to chromatographic resolution.

An alternative mobile phase modifier, difluoroacetic acid (DFA), might also be worth consideration.<sup>1,2</sup> DFA is less acidic and less hydrophobic than TFA and it lowers the surface tension of droplets during electrospray, which confers notable gains in MS sensitivity. Additionally, DFA has been shown to sometimes yield higher resolution separations when used in place of FA and TFA for protein separations.<sup>3</sup> However, it has also been shown that it is necessary to use a purified form of DFA if high quality mass spectra are to be obtained.<sup>3</sup>

This application note expands upon the noted advantages of DFA. With this work, we have explored the use of purified DFA (IonHance DFA) for peptide mapping and found value in its unique selectivity and ability to afford low adduct mass spectra. Moreover, we have compared the performance of DFA peptide mapping against traditional approaches based on FA and TFA.

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## Experimental

### Sample preparation

A reduced and alkylated tryptic digest of NIST Reference Material 8671 (NIST mAb) was acquired in the form of the Waters mAb Tryptic Digest Standard (p/n: 186009126). Each vial of this standard contains approximately 40  $\mu\text{g}$  of peptides and was reconstituted in 80  $\mu\text{L}$  of 0.1% (v/v) formic acid (FA) in 18.2 M $\Omega$  water.

### Method conditions

LC system:	ACQUITY UPLC H-Class Bio coupled to a Xevo G2-XS QToF
Data management:	MassLynx Software v4.1 UNIFI Scientific Information System v1.8
Columns:	ACQUITY UPLC Peptide BEH C <sub>18</sub> , 130 Å, 1.7 $\mu\text{m}$ , 2.1 $\times$ 150 mm (p/n: 186003556) ACQUITY UPLC Peptide CSH C <sub>18</sub> , 130 Å, 1.7 $\mu\text{m}$ , 2.1 $\times$ 150 mm (p/n:186006938)
Mobile phase A:	0.1% FA, DFA, or TFA (v/v) in water
Mobile phase B:	0.1% FA, DFA, or TFA (v/v) in acetonitrile
Column temp.:	80 °C

Detection (UV):	210 nm (separations using FA) 214 nm (separations using TFA) 219 nm (separations using DFA)
Injection volume:	10 $\mu$ L
Gradient:	Initial gradient starts at 1.0% B (BEH column) Initial gradient starts at 0.5% B (CSH column)

Time	Flow	%A	%B	Curve
Initial	0.200	99.0 or 99.5	0.5 or 1.0	Initial
10.00	0.200	99.0 or 99.5	0.5 or 1.0	6
75.00	0.200	60.0	40.0	6
76.00	0.200	20.0	80.0	6
80.00	0.200	20.0	80.0	6
81.00	0.200	99.0 or 99.5	0.5 or 1.0	6
100.00	0.200	99.0 or 99.5	0.5 or 1.0	6

## Data treatment and analysis

Effective peak capacity ( $P_c^*$ ) was calculated from Eq. 1, where  $\Delta t$  is the retention time difference between the HC:T2 and LC:T15 peptides of the tryptic digest standard. Additionally, Eq. 2 was used

to judge the relative abundance of unmodified versus modified peptide species.

$$P_c^* = 1 + \frac{\Delta t}{W_{50\%, \text{avg}}}$$

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*Effective peak capacity equation, where  $\Delta t$  is the difference in retention times for the first eluting peak and the last eluting peak (HC:T2 and LC:T15 peptides of the tryptic digest standard), and  $W_{50\%, \text{avg}}$  is the average peak width at half height for the analyte(s) under investigation.*

$$\text{Relative Abundance} = 100 \times \frac{\sum EIC_2}{\sum EIC_1 + \sum EIC_2}$$

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*Calculation for the relative abundance of a peptide species, where  $\sum EIC_2$  is the summed peak area of the extracted ion chromatogram of the secondary species and  $\sum EIC_1$  is the summed extracted ion chromatogram of the primary species.*

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## Results and Discussion

### Alternative Selectivity

Recent work on the use of IonHance DFA for intact, subunit, and peptide levels of analysis has indicated that DFA is a promising alternative to FA or TFA.<sup>3,4,5</sup> The structures of each acid are shown in Figure 1. The value of DFA as a mobile phase modifier for peptide separations has been previously

reported using an eight-peptide standard.<sup>5,6</sup> To further expand upon its potential in peptide mapping, the chromatographic and MS performance of DFA versus FA and TFA is further reviewed here.

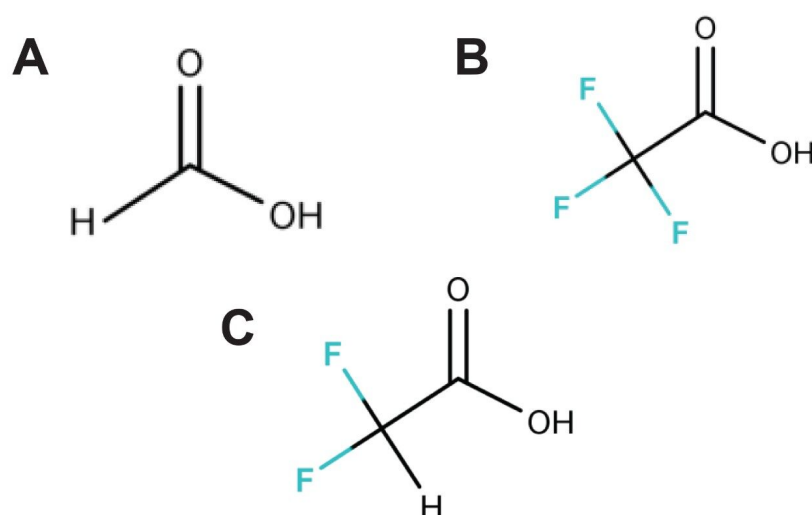


Figure 1. Structures of (A) formic acid (FA), (B) trifluoroacetic acid (TFA), and (C) difluoroacetic acid (DFA).

Key to these experiments has been the availability of IonHance DFA, a purified, MS-grade form of DFA that is certified to contain no more than 100 ppb of sodium and potassium.

Using IonHance DFA as well as MS-grade FA and TFA, we have thoroughly investigated the peptide mapping of a reduced and alkylated tryptic digest of NIST reference material 8671 (NIST mAb). Two hybrid silica columns, an ACQUITY UPLC Peptide BEH C<sub>18</sub> Column and an ACQUITY UPLC Peptide CSH C<sub>18</sub> Column, were chosen for use because of their known capabilities for peptide mapping separations. The CSH particle technology differs from the BEH Technology in that its ligand density is slightly lower, and that it was designed with a low-level surface charge to improve sample loadability and peak asymmetry in low ionic strength mobile phases. This surface charge is the primary contributor to the different selectivities found between the BEH and CSH columns.<sup>7,8</sup>

Figure 2 presents the UV and BPI chromatograms of separations resulting from the BEH column, while Figure 3 depicts chromatograms obtained with the CSH column. Due to the variances in UV absorptivity between the acids, different wavelengths were found to optimize the baseline properties of the separation (Figures 2A and 3A). Additionally, since each stationary phase imparts a unique selectivity and retentivity to the separation, it was necessary to make slight adjustments to

the gradient conditions depending on the employed combination of reagent and column technology.

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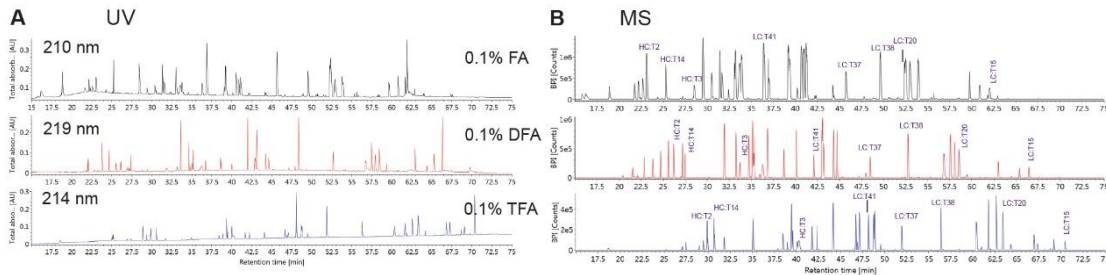


Figure 2. Representative (A) UV and (B) BPI chromatograms from a peptide mapping separation of the mAb Tryptic Digestion Standard using 0.1% FA, IonHance DFA, or TFA-modified mobile phases and an ACQUITY UPLC Peptide BEH  $C_{18}$ , 130 Å, 1.7 μm, 2.1 × 150 mm Column.

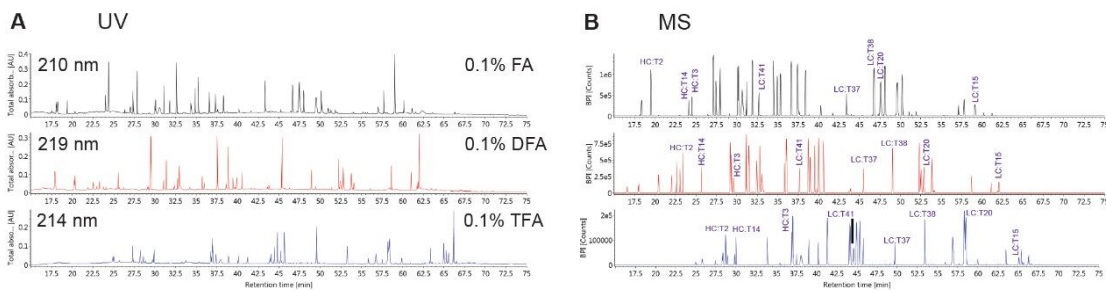


Figure 3. Representative (A) UV and (B) BPI chromatograms from a peptide mapping separation of the mAb Tryptic Digestion Standard using 0.1% FA, IonHance DFA, or TFA-modified mobile phases and an ACQUITY UPLC Peptide CSH  $C_{18}$ , 130 Å, 1.7 μm, 2.1 × 150 mm Column

Differences in selectivity and retentivity can be visualized by comparing separations using FA, DFA, and TFA. Peptide retentivity increases in the order of FA to DFA to TFA, which is due to the increase in ion pairing as the strength and hydrophobicity of the acid increases. More effective ion pairing helps to neutralize the basic sites on peptides or proteins and minimizes secondary interactions, reducing peak tailing and increasing retention. As DFA exhibits properties between FA and TFA, it offers an alternative to their conventional selectivity and retentivity. This could be advantageous to analysts needing to develop certain peptide maps and separations for resolving certain critical peptide species.

## MS Sensitivity and Charge State Distributions



The BPI chromatograms in Figures 2B and 3B indicate that separations using DFA provide slightly lower MS sensitivity than separations using FA but higher sensitivity than separations with TFA. This can be further confirmed through the evaluation of the peak areas of eight peptides, which is shown in Figure 4. Duplicate runs from both the BEH and CSH columns were analyzed and the resulting data were averaged together to generate a comparison plot representing the use of each acid. The peptides were chosen to offer a broad distribution of size and hydrophobicity.

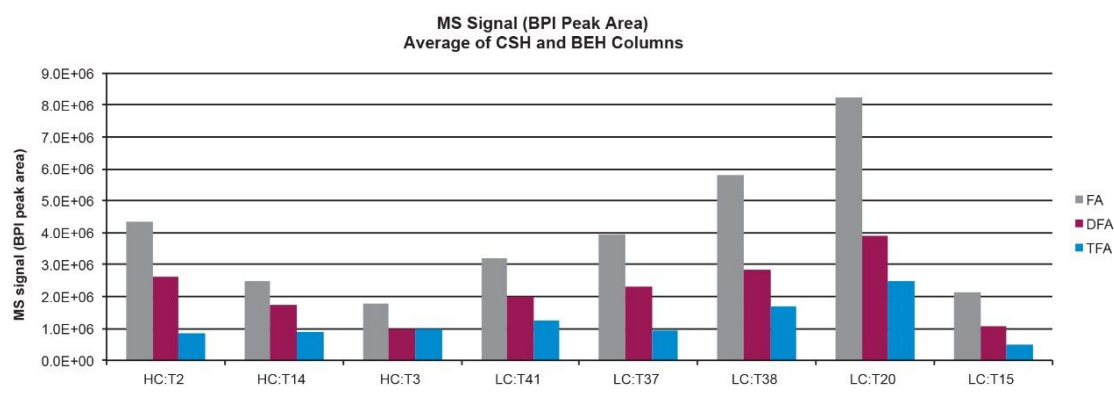


Figure 4. Average peak areas taken from the BPI chromatograms of eight tryptic peptides from the separation of the mAb Tryptic Digestion Standard using 0.1% FA, IonHance DFA, or TFA modified mobile phases and an ACQUITY UPLC Peptide BEH  $C_{18}$ , 130 Å, 1.7  $\mu\text{m}$ , 2.1  $\times$  150 mm Column and ACQUITY UPLC Peptide CSH  $C_{18}$ , 130 Å, 1.7  $\mu\text{m}$ , 2.1  $\times$  150 mm Column.

Table 1 lists the analyzed peptides and their weighted average charge states. The mass spectra corresponding to each data point are shown in Figure 5. Charge states observed with DFA tended to be between those observed with FA and TFA. With subunit separations, similar charge states have been seen between FA and DFA, suggesting that their ionization efficiencies may be more similar than DFA and TFA, at least for high(er) molecular weight regimes.<sup>3</sup> Here, with peptide mapping, the most abundant charge state is generally the same between FA and DFA, with some exceptions.



Peptide	Most abundant m/z	Charge (FA)	Charge (DFA)	Charge (TFA)	Sequence	Modifications
HC:T2	541.269	1.8	1.5	1.7	VTITCSASSR	Carbamidomethyl C
HC:T14	712.7153 1068.488 (TFA)	2.6	2.5	2.2	VDNALQSGNSQESVTEQDSK	
HC:T3	541.3097 811.3977 (TFA)	3.2	2.4	2.3	VGYMHWYQQKPGK	
LC:T41	561.1008 (FA) 701.12171 (DFA) 934.422 (TFA)	4.1	3.4	3.2	WQQGNVFSCSVMHEALHNHYTQK	Carbamidomethyl C
LC:T37	848.7787 1272.5622 (TFA)	2.6	2.5	2.3	GFYPSDIAVEWESNGQPENNYK	
LC:T38	937.4609	2.1	1.9	2.0	TTPPVLDSDGSFFLYSK	
LC:T20	711.9565 948.8188 (TFA)	3.6	3.2	2.9	THTCPPCPAPELLGGPSVFLFPPKPK	Carbamidomethyl C (2)
LC:T15	1343.5933 1679.0841 (TFA)	5.4	5.5	5.3	DYFPEPVTVSWNSGALTSGVHTFPA VLQSSGLYSLSSVVTVPSSSLGTQT YICNVNHKPSNTK	Carbamidomethyl C

Table 1. List of the eight tryptic peptides from the mAb Tryptic Digestion Standard analyzed for peak area and peak capacity, where modification sites are labeled in red. Analyses were performed using 0.1% FA, IonHance DFA, or TF-modified mobile phases and an ACQUITY UPLC Peptide BEH C<sub>18</sub>, 130 Å, 1.7 μm, 2.1 × 150 mm Column and ACQUITY UPLC Peptide CSH C<sub>18</sub>, 130 Å, 1.7 μm, 2.1 × 150 mm Column. Charge states are reported based on weighted averages of observed ion intensities.

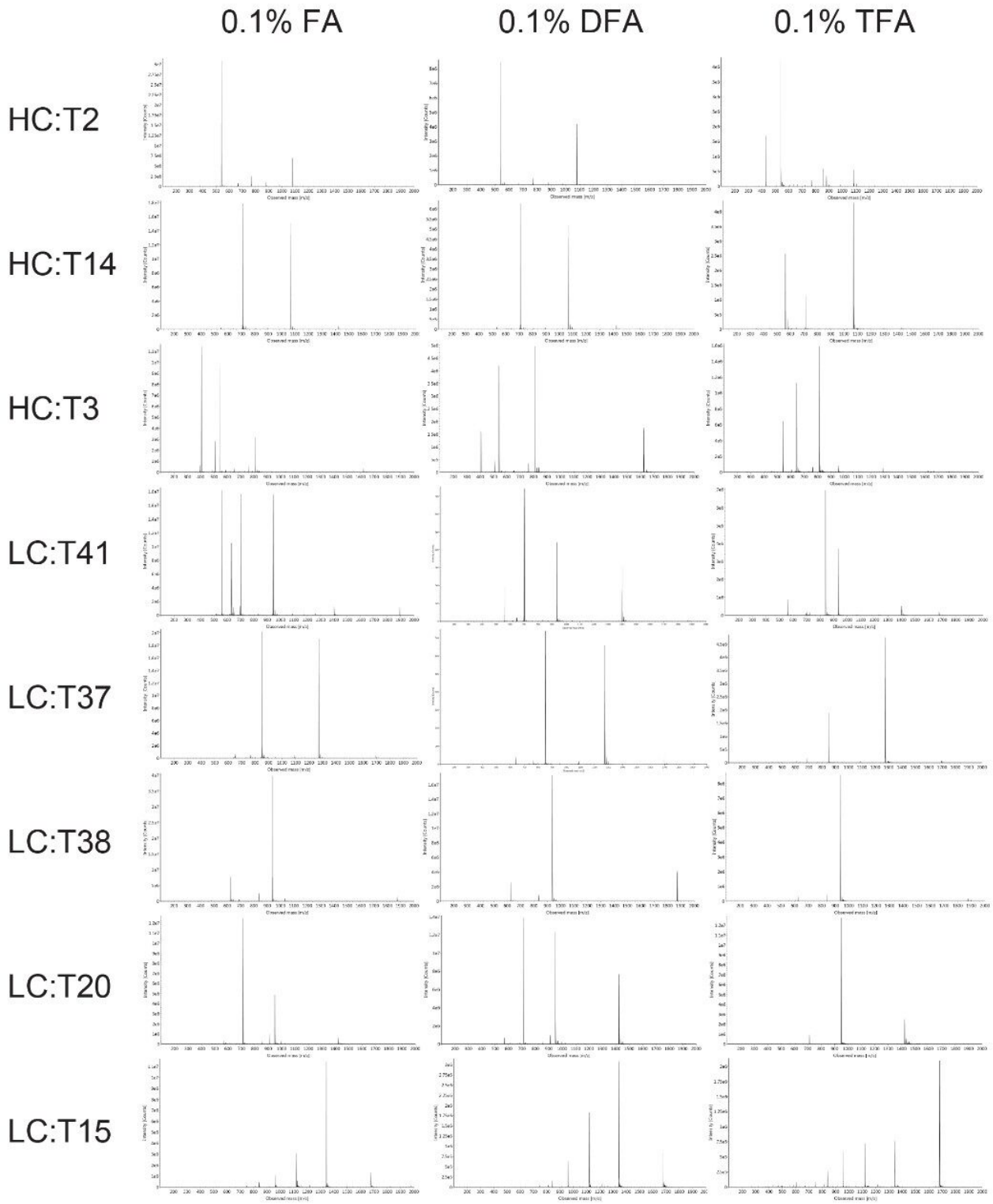


Figure 5. Mass spectra of the eight tryptic peptides from the mAb Tryptic Digestion Standard analyzed for peak area and peak capacity. Analyses were performed using 0.1% FA, IonHance DFA, or TFA-modified

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*mobile phases and an ACQUITY UPLC Peptide BEH C<sub>18</sub>, 130 Å, 1.7 μm, 2.1 × 150 mm Column.*

An example of this can be seen in the mass spectra of the HC:T14 peptide that are shown in Figure 6. As can be seen, charge state abundances differ significantly with separations using TFA versus FA and DFA. Here, the relative abundances of the 3+ and 2+ charge states are very similar with FA and DFA, but when using TFA, the 2+ charge state becomes much more prominent. For all but one of the analyzed peptides, FA and DFA share the same most abundant charge state. However, in terms of weighted average charge state, DFA can sometimes yield charge states closer to those seen with TFA. An example is the early eluting peptides of NIST mAb. Interestingly, many of these peptides show charge states that are lowest using DFA followed by TFA, as exemplified by the HC:T2 peptide in Figure 5. This suggests that FA, DFA, and TFA may differ in how their charge state distributions are affected by certain characteristics of peptides or gradient conditions.

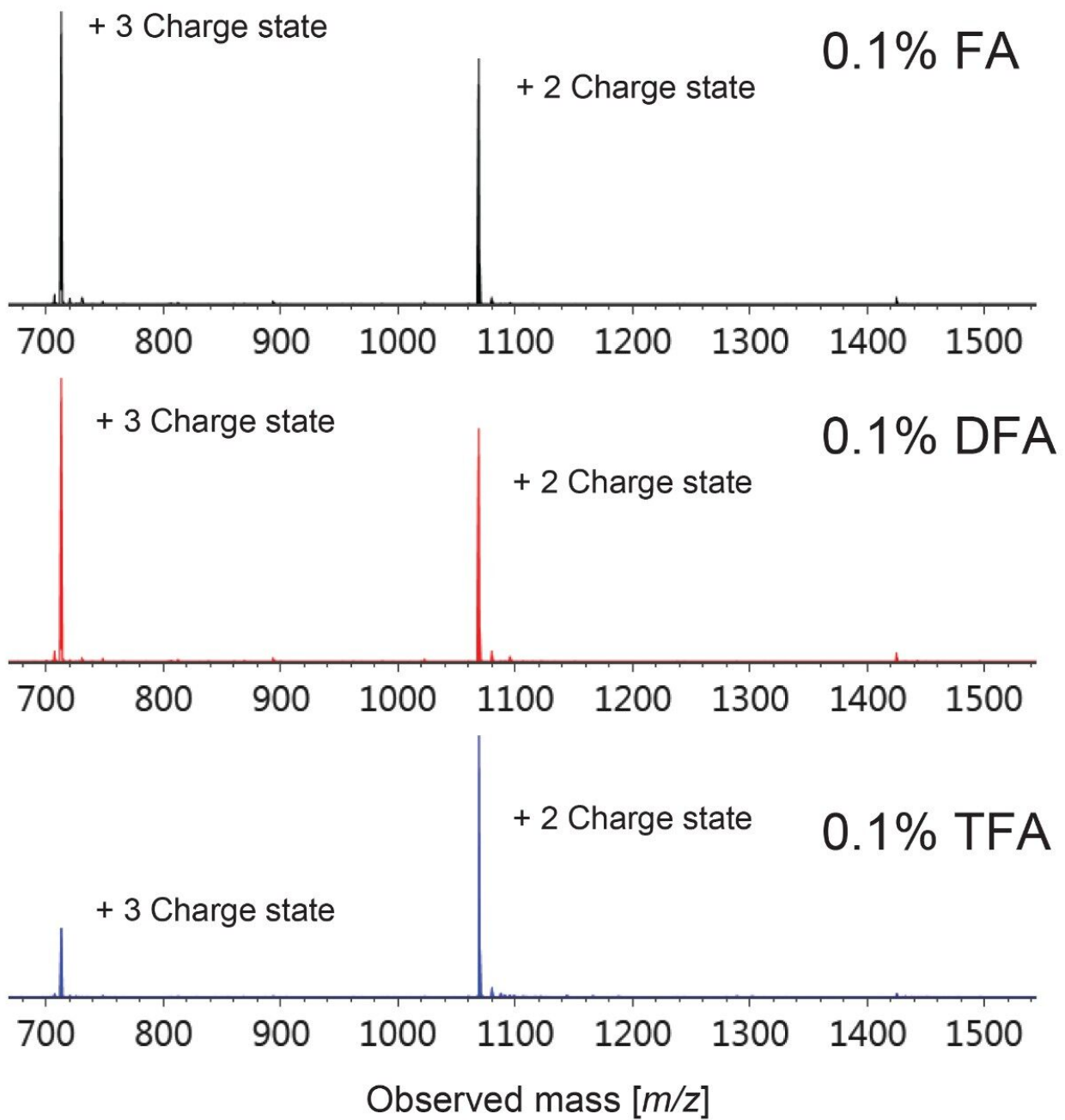


Figure 6. Mass spectra of the HC:T14 peptide from a peptide mapping separation of the mAb Tryptic Digestion Standard using 0.1% FA, IonHance DFA, or TFA-modified mobile phases and an ACQUITY UPLC Peptide BEH  $C_{18}$ , 130 Å, 1.7  $\mu$ m, 2.1  $\times$  150 mm Column.

Finally, the properties of peptides containing modifications were also evaluated between each acid modifier. An example of one type of modified peptide is shown in Figure 7. The peptide under review is LC:T37 and it was found that it has a deamidated counterpart. To estimate the relative abundance

of the deamidated species to the primary species, the peak areas from the extracted ion chromatograms of both were taken. For separations using FA or DFA, the relative abundances of deamidated LC:T37 peptide were calculated to be 2.6% while for separations with TFA, it was 2.5%. The similarity in these results suggests that each method is capable of yielding the relative abundances of modified peptide species.

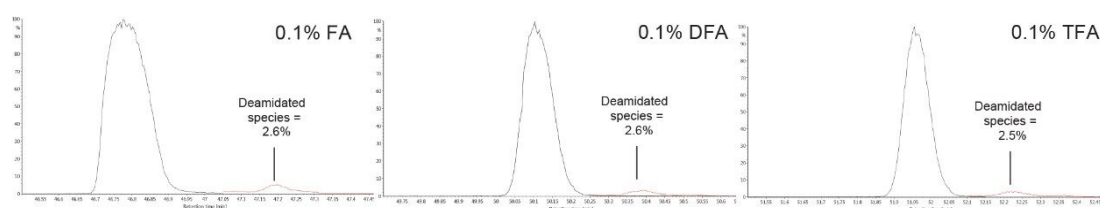


Figure 7. Extracted ion chromatograms of the unmodified and deamidated species of LC:T37 from the mAb Tryptic Digestion Standard using 0.1% FA, IonHance DFA, or TFA-modified mobile phases and an ACQUITY UPLC Peptide BEH C<sub>18</sub>, 130 Å, 1.7 μm, 2.1 × 150 mm Column.

## Resolution

When assessing LC resolution, we found that DFA offers high peak capacities nearer those obtained with TFA (Figure 8). These peak capacities were calculated from Eq. 1 with peak widths at half height taken from the extracted ion chromatograms of the eight peptides.

The results in Figure 8 fall into place in a predictable way. The strongest acid and most hydrophobic ion pairing reagent, TFA, provided the highest resolution while the weakest and least hydrophobic, FA, gave the lowest. It can also be seen that, for this separation, the CSH column produced higher peak capacities than the BEH column. This is likely due to favorable contributions from its positively charged stationary phase. However, it is worth noting that separations combining DFA or TFA with the BEH column can show higher resolution than the CSH column coupled with FA modified mobile phases.

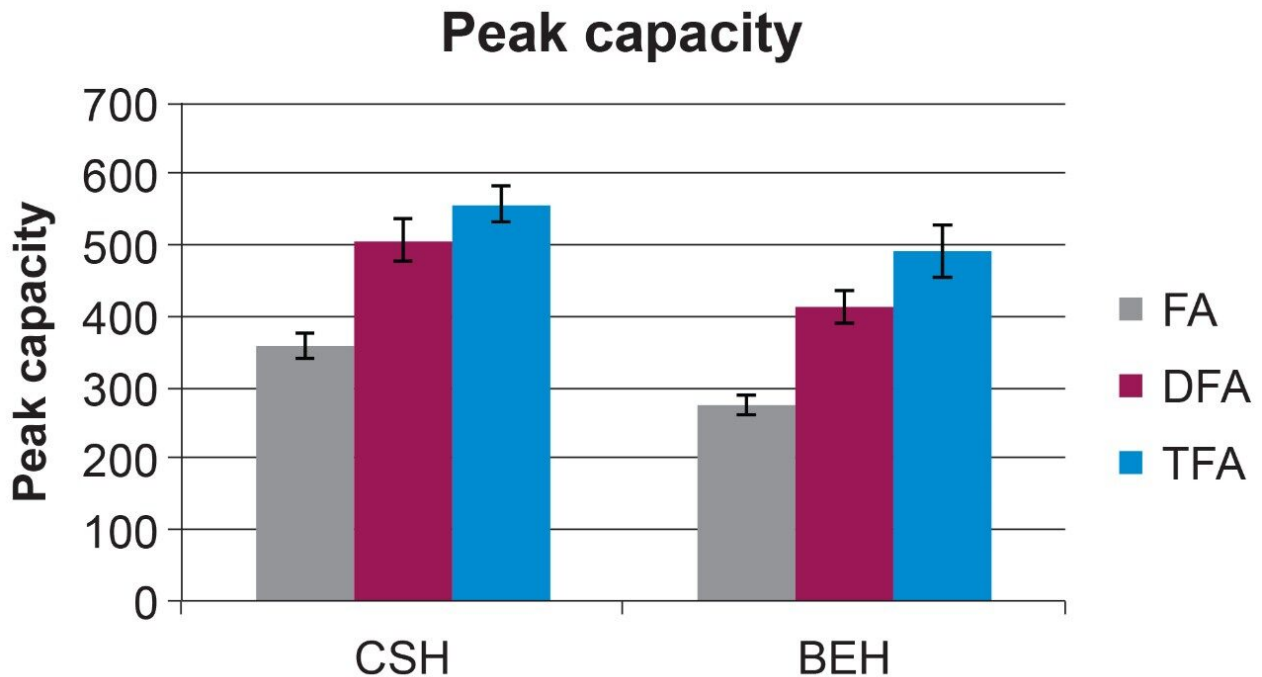


Figure 8. Effective peak capacities based on eight tryptic peptides from the mAb Tryptic Digestion Standard. Analyses were performed using 0.1% FA, IonHance DFA, or TFA-modified mobile phases and either an ACQUITY UPLC Peptide BEH C<sub>18</sub>, 130 Å, 1.7 µm, 2.1 × 150 mm Column or ACQUITY UPLC Peptide CSH C<sub>18</sub>, 130 Å, 1.7 µm, 2.1 × 150 mm Column. Reported averages are based on duplicate analyses.

Yet another important observation is one related to the distribution of peptides across the chromatogram and corresponding separation space. Although FA separations show comparatively low peak capacities, they appear to give the most appealing distribution of peptides across the chromatograms, with somewhat fewer clusters of peptides. TFA seems to produce the poorest distribution. This distribution and apparent selectivity can be advantageously exploited when needing to separate particular groups or characteristics of peptides.

### Low Adduct MS Spectra

Mobile phase quality will often correlate with mass spectral quality. This can be demonstrated through the use of commercial, reagent quality DFA versus MS grade FA and IonHance DFA, which has been purified to MS-grade specifications.<sup>6</sup> Figure 9A indicates that while reagent grade DFA can give similar chromatography to purified DFA, it contains a high concentration of sodium that can readily complicate MS spectra with sodiated adducts (Figure 9B). Trace-level metal contaminants generally do not affect protein or peptide separations, but they can disrupt the interpretability of the

mass spectra by distorting relative abundances of protonated species and causing spectral crowding. Using a reagent certified to have low metal content, such as IonHance DFA, it is possible to acquire high quality MS data.

Interestingly, a difference can be found when comparing the sensitivity of peptides versus protein subunits to the formation of sodium adducts. As exemplified in Figure 9B, peptide mass spectra can be populated with sodiated ions corresponding to as much as 20% of the main peak when an inferior quality of acid modifier is employed. In contrast, the sodiated ion of the light chain from NIST mAb tends to only approach a level of 4%.<sup>3</sup> This suggests that certain analyte classes and properties might be more susceptible than others to the formation of ion adducts.

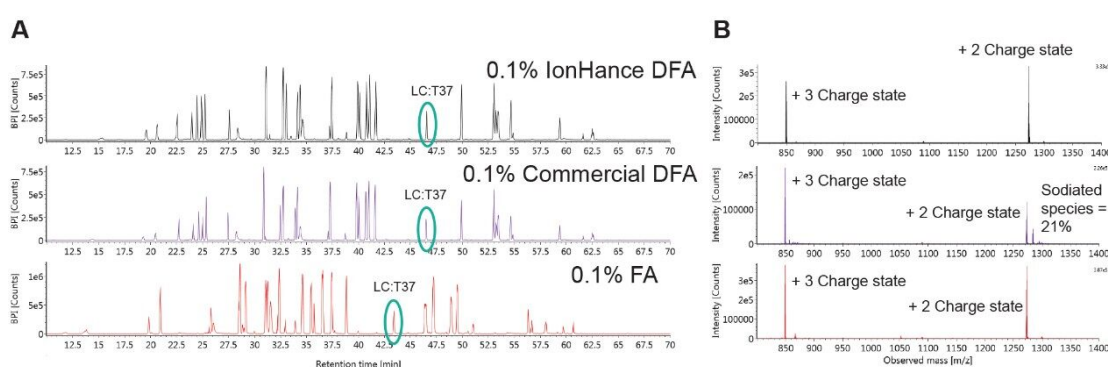


Figure 9. (A) BPI chromatograms and (B) mass spectra of the LC:T37 peptide from a peptide mapping separation of the mAb Tryptic Digestion Standard using 0.1% IonHance DFA, reagent grade DFA, and FA-modified mobile phases and an ACQUITY UPLC Peptide CSH C<sub>18</sub>, 130 Å, 1.7 μm, 2.1 × 150 mm Column.

## Conclusion

Peptide mapping separations are typically run using FA- or TFA-modified mobile phases, with FA providing better MS sensitivity than TFA and TFA producing higher peak capacities than FA. However, an alternative mobile phase modifier, DFA, has been developed that can give a better balance between chromatographic and mass spectrometric performance. IonHance DFA, a stringently purified form of DFA containing less than 100 ppb of sodium and potassium, offers unique selectivity for peptide RPLC separations while also giving strong MS sensitivity gains over TFA and improved



peak capacities over FA. Its low metal content is essential to minimizing adduct formation in peptide mass spectra. Ultimately, IonHance DFA provides analysts with a valuable ion pairing option for improving LC-MS separations of peptides given that it confers unique selectivity and sensitivity gains that can prove advantageous for method development.

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